B. Sc. (Physics) Semester – V

Paper - 503 Sub Code 4296: Atomic Physics and Nuclear Instrumentation

Tin 2:30 I		Nov=2014	[Total Marks70
Instructions:	1. Symbols have their	usual meaning.	
	2. Figures on right han	d side show marks of that qu	uestion.
Q:1:	~ 1 a		
(a) Explain the Stark effect with experimental setup.		[10]	
	_	J are obtained from the com	nponents
of spectral	term.		[04]
		OR	
Q:1: (a) Draw the 6	energy transition diagrar	n of ${}^2P_{3/2} \rightarrow {}^2S_{1/2}$ line of Ano.	malous
Zeeman effect with tabulation.			[10]
(b) Calculate	the change in energy for	the sodium source emitting	the light of
ယ့် a velengt	th 5890 Å in the presenc	e of magnetic field of 1T str	ength. [04]
Q:2:	1 D 1 C (2D		· · ·
• •		ive an expression for change	
an example of principal series of lithium.		[09]	
(b) Calculate	Lande's "g" splitting fac	stor for the terms $^{10}\mathrm{H}_{1/2}$ and 1	$^{10}G_{1/2}$ [05]
		OR	
Q:2: (a) Discuss ah	out the Orbital and Spin	magnetic moment of an ato	om. [07]
	_	of Lande's "g" splitting fact	
. , 1	-	- 1	. J
Q:3: (a) Derive We	eizsacher's semi-empiric	al Binding Energy formula i	in the case of liquid
drop mode	el.		[10]
(b) Write a sh	ort on binding energies	of nuclei.	[04]

OR

0.2.			
Q:3: (a) Explain magic numbers for protons with diagram. (b) Explain in short: Predictions of the Shell model.			
			Q:4: (a) Discuss evidences that support the existence of magic numbers.
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(b) Write a note on Standard model of elementary particles.	[07]		
OR			
O:4:			
(a) Discuss the fundamental interactions and conservation laws in Physics.			
(b) Justify why neutron possess little magnetic moment though it doesn't have net charge	e.[06]		
Q:5:			
(a) Explain in detail the technique of NMR.			
(b) Explain applications of NMR in chemistry.			
OR			
Q:5:	[08]		
(a) What is Mossbauer effect? Explain the Mossbauer effect with experimental setup.			
(b) Explain application of Mossbauer effect for precise energy measurement.	[06]		
XXX			