

- Q-1(a) Give an account of reactions mechanism of chain reaction between hydrogen and bromine molecules. 08
- (b) State the characteristics of chain reaction. 06
- OR
- Q-1(a) Explain auto oxidation. Give an account of kinetic of hydrogen and oxygen molecules. 08
- (b) Show that the decomposition of Ozone follows first order kinetics. 06
- Q-2(a) Derive an expression for Fermi-Dirac distribution law. 10
- (b) Calculate the rotational partition function of hydrogen gas at 0° C. 04
 $I = 0.459 \times 10^{-40} \text{ gm cm}^2$, $\sigma = 2$
- OR
- Q-2(a) Derive an expression for Bose-Einstein Distribution Law and hence Boltzmann Distribution Law. 10
- (b) Calculate the translation partition function for 100 cm³ of hydrogen gas at 25°C. 04
- Q-3(a) Derive the general expression for E M F of reversible cells. 07
- (b) The standard potential of the Ag, Ag₂O(s), OH⁻ electrode is -0.344 Volt at 25°C. The heat of formation of silver oxide is -7.300 cal at 25°C and ΔC_p is about 1.0 cal deg⁻¹ mole⁻¹. Estimate the temperature at which silver oxide will dissociate freely in air. 07
- OR
- Q-3(a) Discuss the determination of dissociation constant by E M F method. 07
- (b) The E.M.F. of the cell Zn/ZnCl₂ (1.0 m)/AgCl(s), Ag is 1.015 Volts at 0°C and 1.005 Volts at 25°C. assuming the temperature co-efficient to be approximately constant in this vicinity calculate the heat change of the cell reaction at 25°C. 07
- Q-4(a) State different types of partition functions and factors affecting them. 08
- (b) Define: Chain length, Degree of polymerization, Partition function Polycondensation, Thermodynamic probability and Ensemble. 06
- OR
- Q-4(a) Discuss the kinetics and mechanism of Ring Scission Polymerization. 08
- (b) Explain the difference between: 06
- (i) Macro and Microstate
- (ii) Irreversible and Reversible electrodes
- Q-5 Write short notes on: (Any Three) 14
- (i) Stepwise Polymerization
- (ii) Ionization constant of water
- (iii) Ring scission polymerization
- (iv) Equilibrium constant of metathetic reactions
- (v) Single electrode potential